## **MOLE WORKSHEET #1**

- 1. How many moles are there in 143 bananas?
- 2. How many moles of bicycles are there in 63 bicycles?
- 3. How many cars are there in 2.5 moles of cars?
- 4. How many wheels are there in 2.6 moles of bicycles?
- 5. How many moles of wheels are there in 4.5 moles of cars?
- 6. How many molecules are there in 3.6 moles of molecules?
- 7. How many moles of molecules are there in 4.3 x 10<sup>23</sup> molecules?
- 8. How many atoms are there in one molecule of aluminum bromide?
- 9. How many atoms are there in 43 molecules of calcium sulphate?
- 10. How many moles of atoms are there in 3.8 moles of water molecules?
- 11. How many molecules of hydrogen sulfate can be produced from 4.3 moles of hydrogen atoms?
- 12. How many oxygen atoms are there in 3.8 moles of water?

## **PART B**

- 13. What is the mass of one mole of water?
- 14. What is the GMM of lead II acetate?
- 15. How many moles are there in 43 g of sodium atoms?
- 16. How many moles are there in 146 g of hydrogen?
- 17. What is the mass of 6.5 moles of copper atoms?
- 18. What is the mass of 4.8 moles of silver atoms?
- 19. What is the atomic mass of element XX if 3.6 moles of the substance has a mass of 192 grams?
- 20. How many moles are there in 48 grams of water?
- 21. How many grams are there in 0.361 moles of nitric acid?
- 22. What is the GMM of a substance if 43 grams is the mass of 1.4 moles?
- 23. What is the mass of 4.50 moles of magnesium chloride?
- 24. What is the mass of 0.0630 moles of potassium phosphate?
- 25. What is the mass of 0.0630 x 10<sup>-4</sup> moles of aluminum sulphate?
- 26. What is the number of moles in 6.6 grams of silicon tetrachloride?
- 27. How many atoms are there in 43 molecules of hydrogen sulphate?

## Answers:

- 1. 2.38 X 10 <sup>-22</sup> mol
- 2. 1.0 X 10 <sup>-22</sup> mol 3. 1.5 X 10 <sup>24</sup> cars 4. 3.1 X 10 <sup>24</sup> wheels

- 5. 18 mols
- 6. 2.2 X 10 24 molecules
- 7. 7.1 X 10<sup>-1</sup> mol
- 8. 4 atoms
- 9. 258 atoms

- 10. 11 mols (11.4 mols) 11. 1.3 X 10 <sup>24</sup> molecules 12. 2.3 X 10 <sup>24</sup> oxygen atoms
- 13. 18 g
- 14. 325.2 g
- 15. 1.9 mol
- 16. 73 mol
- 17. 410 g (412.8 g)
- 18. 520 g (517.9 g)
- 19. 53 g (53.3 g)
- 20. 2.7 mol
- 21. 22.7 g
- 22. 31 g/mol
- 23. 429 g
- 24. 13.4 g
- 25. 2.16 X 10<sup>-3</sup> g 26. 3.9 X 10<sup>-2</sup> mol
- 27. 301 atoms